

Potential food additive of *Boswellia carterii* essential oil encapsulated within gum arabic: a particle size distribution and zeta potential analysis

¹Barre, M.S., ^{1,*}Ali, F.B., ²Mirghani, M.E.S., ²Hazri, N.F., ³Anuar, H. and ³Nasir, N.A.M.

¹Department of Biotechnology Engineering, Kulliyah of Engineering, International Islamic University Malaysia (IIUM) Gombak Campus, P. O Box 10, 50728 KL, Malaysia

²International Institute for Halal Research and Training, IIUM Gombak Campus. P. O Box 10, 50728 KL, Malaysia

³Department of Manufacturing and Materials Engineering, Kulliyah of Engineering, International Islamic University Malaysia (IIUM) Gombak Campus, P. O Box 10, 50728 KL, Malaysia

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Abstract

Boswellia carterii (BC) {*Burseraceae* family} essential oil (EO) were extracted by hydrodistillation process. Gum Arabic (GA) {*Acacia senegal*} polymer particles containing a BCEO were prepared by spray drying technique. The mean particle size and its distribution, as well as the zeta-potential of the microcapsules, were analyzed and found Z-Average $382\pm 203\text{nm}$, PDI 0.77 ± 0.3 , ZP- $25\pm 2.73\text{mV}$, respectively. Product encapsulation efficiency (EE %) was found at $75\pm 0.8\%$. The surface morphology of the particles was obtained by scanning electron microscope (SEM). Furthermore, particles moisture content was analyzed by the oven drying method. The efficiency of encapsulation (EE %) was estimated by specifying the content of essential oil in the product. Gas chromatography (GC) coupled with time-of-flight mass spectrometry (TOF-MS) analysis of EO has been performed to determine the chemical compounds and their prevalence concentrations respectively. The composition of initial essential oil (added in the emulsion) and the encapsulated essential oil (extracted from spray dried powder) were analyzed and compared. The outcome of the research encourages the high potentiality and usefulness of the product in the food industries sector as a food additive agent, moreover, it suggests for further research to unravel potential implementation of BCEO microcapsules in the food production chain

1. Introduction

Essential oils are liquid secondary metabolites which synthesized by various organs of aromatic plants such as “leaves, flowers, rhizomes, seeds, stems, and plant exudates like gums and resins” (Andrade *et al.*, 2014) and they have many interesting applications in Food, Cosmetic, and Pharmaceutical industries (Rao *et al.*, 2019). However, these oils are highly concentrated volatile substances which are very sensitive to the surrounding environmental conditions “temperature, humidity, and oxygen” (Burt, 2004; Dubey *et al.*, 2009; Kotronia *et al.*, 2017).

Essential oils (EOs) had a tremendous shining history in the ancient times for their useful advantages, such as their nutraceutical properties and have gained a renewed interest recently because of their widely diversified applications and numerous relevant biological activities, which makes them more useful in different

sects of industries (Elshafie and Camele, 2017). EOs are mostly extracted from several medicinal and aromatic plants by distillation method (Reyes-Jurado *et al.*, 2014; Cui *et al.*, 2017; Guan *et al.*, 2019). Usually, they possess antioxidant, antimicrobial properties and used as food preservatives (Fathi *et al.*, 2014; Preedy, 2016).

Hence, recently in food industries alone, many studies are showing the benefit of these natural products and their potentiality to replace artificial food additives such as flavoring and preserving agents (Fathi *et al.*, 2014; Basak and Guha, 2018). Other industries also (pharmaceutical, cosmetic, textile) use EOs into their value-added products, which enhances and improves the consumer habit of adopting these newly improvised formulations (Carvalho *et al.*, 2015; Sharifi-Rad *et al.*, 2016).

Hence, to obtain these valuable volatile liquids from the plant matrix, different methods had been employed

*Corresponding author.

Email: fathilah@iium.edu.my

(Fahim *et al.*, 2017). The extraction methods which had been used to get EOs from the plants kept on changing consistently over time. However, the most frequently used methods were; hydrodistillation, steam-distillation, organic solvent extraction, furthermore, recently have been added view more extraction techniques, such as; supercritical extraction, microwave extraction, ultrasonic extraction, Ohmic assisted extraction etc, depending on the plant species and plant organs to be extracted (Fahim *et al.*, 2017; Gavahian and Farahnaky, 2018; Turk *et al.*, 2018; Giacometti *et al.*, 2018).

In the present study gum resin from BC were extracted by hydrodistillation process. The EO obtained was analyzed by GC-TOF-MS to determine its chemical composition and the predominant compounds. Hence, to get the full benefit of them, encapsulation techniques ensures the protection of the fragile oils and their controlled release to the desired target (Ahn *et al.*, 2008). Moreover, it enhances the versatile formulations and diversified applications. Proper selection of coating materials and their concentration ratios is of the utmost importance for spray drying to obtain the required functional characteristics for the final product of microparticles.

Hence, the most commonly used carriers (coating/wall materials) in micro and nanoencapsulation processes are polymers, gelatin, liposome and solid lipid nanoparticles (Wang, 2015). Thus, the main objectives in the present study summarized as follows; extraction of EO from BC gum resin by Hydro-distillation process and emulsifying them with Gum Arabic polymer particles as a carrier agent using deionized water. After homogenizing the emulsion properly, the spray drying process was performed to dry it and embedded EO into the Gum-Arabic polymer, subsequently, particle analysis and characterization were performed to determine their stability and size distribution.

2. Materials and methods

2.1 Materials

BC gum resin was obtained from Somaliland harvesters and identified botanically by botany department of Burco University, (Burco), whereas, gum Arabic (Acacia Senegal) purchased from Perfect Life Company (UAE). The primary solvent used in the experimental process was deionized water (DI W).

2.2 BC Essential oil extraction

To extract the essential oil from BC gum resin hydrodistillation process was employed. Clevenger type of distillation apparatus was used. Ratios 1:10 of gum to DI water were used in the distillation process. After

separation of the EO from hydrosol, it was dried over anhydrous sodium sulfate to eliminate traces of moisture and stored in sealed vials at low temperature (4°C) for further analysis. About 4±0.3% yield of EO by weight was obtained from the process. The Essential oil was analyzed by Gas chromatography (GC) coupled with time-of-flight mass spectrometry (TOF-MS) to identify their chemical constituents. The electron impact (70eV) GC-MS measurement were performed with a Hewlett Packard 5890 II GC equipped with a HP-5 MS capillary column (30m X 0.25mm i.d. 0.25µm) and HP 5972 mass selective detector. A temperature was programmed to start initially from 80°C, followed by the increase of 10⁰C/min until reached up to 270°C with two minutes solvent delay by injection. The injector temperature was adjusted to 200°C for the analysis (Basar, 2005).

2.3 Emulsion preparation

Oil in water (O/W) emulsion preparation procedure is used, and a ratio of the different agents was EO 3g: GA 27g: DI W 70g, respectively. Overnight mechanical agitation was used to homogenize the solution for better emulsion preparation.

2.4 Spray drying process and particle analysis

The BC EO and GA emulsion were dried by BUCHI mini spray dryer B-290. After obtaining the microencapsulated particles different characterization analyses were performed, namely: Particle size average (Z-Aver), Particle size distribution (PDI), and Zeta Potential (ZP) of the particles to determine their stability and interaction properties in the solution. Hence, flowing methods have been used:

- a) For particle size distribution, Zeta potential, and polydispersity index determination, Zetasizer Ver. 7.01; manufactured by MALVERN, was used. System temperature was 25⁰C, Count Rate (kcps) was 135.7, and Cell description; Disposable sizing cuvette. The solution was sieved by Whatman No. 1 before analyzing it.
- b) JOEL 6360LA scanning electron microscope (SEM) JEOL Ltd., Tokyo, JAPAN was used for elucidating the physical structure of BC EO microencapsulated in GA (Figure 2).

3. Results and discussion

The hydrodistillation process of extraction used here produced a considerably high yield (4±0.3% by weight) of EO, and their chemical properties were shown in (Table 1). The average standard yield usually obtained from this species is recorded to be 2.5% to 5% by weight, depending on various factors, including

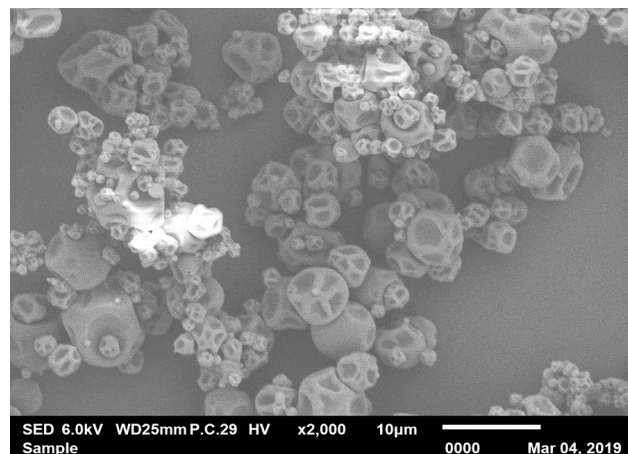
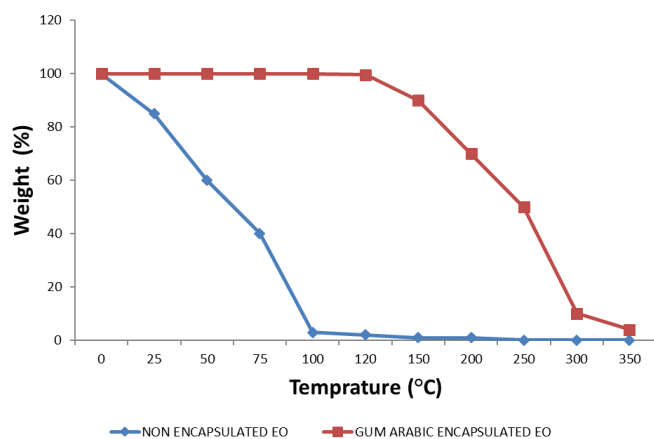


Figure 1. Thermo-gravimetric analysis (TGA,) of encapsulated BC EO with GA and non-encapsulated BC EO

Figure 2. Scanning Electron Microscopy (SEM) analysis of particles

Table 1. Most abundant chemical compounds of BC EO before encapsulation

No	Compound name	Similarity	Area %	Cas No.
1	(-)-Myrtenol	827	1.7844	19894-97-4
2	Naphthalene, 1,2,3,5,6,8a-hexahydro-4,7-dimethyl-1-(1-methylethyl)-, (1S-cis)-	856	4.2162	483-76-1
3	Carveol	860	12.622	99-48-9
4	Cyclohexene, 1-methyl-4-(1-methylethylidene)-	865	1.1363	586-62-9
5	ç-Terpinene	865	11.218	99-85-4
6	à-copaene	867	3.1426	0-00-0
7	Bicyclo[3.1.1]hept-3-en-2-one, 4,6,6-trimethyl-, (1S)-	871	1.8076	1196-01-6
8	à-Phellandrene	912	0.10367	99-83-2
9	1,3,6-Octatriene, 3,7-dimethyl-, (Z)-	958	7.1806	3338-55-4

Table 2. Most abundant chemical compounds of BC EO after encapsulation

No	Compound name	Similarity	Area %	Cas No.
1	Aromandendrene	872	7.2737	539-52-6
2	Propanoic acid, 2-hydroxy-2-methyl-, ethyl ester	883	9.01E-05	123-35-3
3	à-Myrcene	887	8.0539	98-55-5
4	à-Terpineol	895	0.51593	470-82-6
5	Eucalyptol	902	1.6146	80-56-8
6	à-Pinene	907	4.485	3387-41-5
7	Bicyclo[3.1.0]hexane, 4-methylene-1-(1-methylethyl)-	908	9.8309	3779-61-1
8	Cyclohexene, 1-methyl-4-(1-methylethenyl)-, (S)-	929	7.9913	3387-41-5
9	Bicyclo[3.1.0]hexane, 4-methylene-1-(1-methylethyl)-	934	9.8309	127-91-3
10	à-Pinene	962	1.9445	535-77-3
11	Benzene, 1-methyl-3-(1-methylethyl)-	962	4.9508	535-77-3

(harvesting time, storage process, and method of extraction).

The encapsulated particles obtained from the spray drying process have been analyzed from different aspects. Their TGA analysis (Figure 1) shows that the GA gave good protection to the EO and only started to degrade after 150°C of temperature, conversely, non-encapsulated EO started evading at ambient temperature and totally evaporated at 100°C of temperature). GC-TOF-MS analysis of the EO before and after encapsulation shows no significant difference been detected in terms of chemical compounds (Table 1 and 2).

Size, zeta potential, and polydispersity index (PDI) of particles are considered important parameters for the characterization of particles. The Z-Average of particle size was almost 382 ± 203 nm, whereas, the zeta potential was found to be -25.9 ± 2.73 mV, which indicates the good stability of the particles. However, the PDI shows a broadly dispersed particle size (PDI = 0.77). Scanning electron microscope (SEM) as presented in Figure 2 shows semi-spherical tapered particle shape. Hence, optimizing of certain parameters such as the ratio of EO to GA polymer particles in the emulsion preparation could improve the particle size homogeneity and their physical structure also. Maqbool *et al.* (2011) analyzed

the postharvest application of GA with and without Cinnamon and Lemon Grass EOs on some tropical fruits as a shelf life enhancer, they concluded that GA with EOs gave fruits high protection and prolonged their storage period, which is supporting the present research findings. Nurdin *et al.* (2019) used GA as a surfactant on the stability of water-based alumina nanoparticle suspension and concluded that when the concentration of GA increased (from 0.1% to 0.5%) the zeta potential falls down (from -25mV to -36mV). Hence, this could be explained that GA minimizes interaction between particles which in turn prevents the suspension from agglomeration.

4. Conclusion

BCEO is very volatile and highly sensitive to environmental conditions. However, after encapsulation in GA polymer, it becomes more stable and easily transportable, formulation and dosing also been enhanced, which enables the different industries (Food, Cosmetics, and Pharmaceutical) to incorporate it within their value-added products for versatile applications. Hence, to improve particle size distribution stability and physical structure of the particles, optimization of the experimental process conditions is needed, considering the GA and EO ratios important factor in the process.

Conflict of Interest

The authors have no conflict of interest to declare.

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