

Impact of different thin layer drying temperatures on the drying time and quality of butterfly pea flowers

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Abstract

With attractive flower colours ranging from dark green to purple, Butterfly pea (*Clitoria ternatea* L.) is grown year-round in Vietnam. The purpose of this study is to determine the effect of air temperature on drying time and antioxidant compounds of Butterfly pea flowers, fitting the drying curves and testing the goodness of fit. In this study, air drying characteristics of the Butterfly pea flowers were determined using drying air temperature from 55°C to 70°C at a constant air velocity of 1 m/s. The data of experimental moisture loss were fitted to selected seven thin-layer drying models. The effect of drying conditions on the anthocyanin and total phenolic compound changes of Butterfly pea flower were compared. The effect of temperature on the diffusivity was described using the Arrhenius equation with an activation energy of 71.63 kJ.mol⁻¹. At increasing temperature, the effective moisture diffusivity values ranged from 2.39×10⁻¹² and 7.76×10⁻¹² m²s⁻¹. The mathematical models were compared according to the three statistical parameters such as the coefficient of determination (R²), reduced chi-square (χ²) and root mean square error (RMSE) between the observed and predicted moisture ratios. The highest value of R² (99.8%) and the lowest values of χ² (0.0004) and RMSE (0.0178) were observed for drying air temperature of 70°C. Among the seven mathematical models tested with experimental data, the Page model could sufficiently be described the drying characteristics of the Butterfly pea flower.

1. Introduction

Natural colourings are extracted from vegetables, fruits, and tubers, which are available in nature. They do not only offer attractive appearances, but they have also been recognized to increase the nutritional value of the foods. In particular, butterfly pea species (*Clitoria ternatea* L.) possesses beautiful flowers with dark blue to purple petals that can bloom almost year-round in Vietnam. Recent concerns about the use of synthetic food colours and their effects on human health have increased, therefore, natural colourants such as butterfly pea flowers extract have become alternatives commonly in food processing. It has the potential to be used as a natural food colourant in cooking, baking and beverages (Bhowmik *et al.*, 2010) as well as in traditional medicine (Mehmood *et al.*, 2019). The butterfly pea flowers mostly contain alkaloids, flavonoids, taraxerols,

taraxerones, triterpenoids and anthocyanins as active chemicals that bring about its biological effects (Kosai *et al.*, 2015). Delphinidin has a higher antioxidant activity compared with some other anthocyanidins (Marpaung *et al.*, 2013). Its extracts possess a wide range of pharmacological activities including antibacterial, antidiabetic, antidiarrheal, antifungal, anthelmintic, anti-inflammatory, antimicrobial, antioxidant, antidepressant, and antipyretic activities, hypolipidemia, immunomodulatory, and wound healing (Kosai *et al.*, 2015).

Drying is one of the methods that is widely used to preserve fruits and vegetables, in which thin-layer drying is a commonly used method for determining the drying kinetics of fruits and vegetables. Drying significantly reduces volume and weight, minimize packaging, storage and transport costs. It allowed the dried product to be

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stored easily in ambient conditions. Thus, the selection of the most suitable thin-layer drying model is also a very important tool in describing the drying behaviour of fresh fruits, vegetables and edible flowers. It was observed that butterfly pea flowers are grown year-round but they are highly perishable and therefore require careful treatment in order to preserve the bioactive compounds, especially total polyphenol and anthocyanins. Numerous experimental studies and mathematical models on the drying characteristics of various fruits and vegetables have been carried out, such as papaya (Lemus-Mondaca *et al.*, 2013), carrot (Kaya *et al.*, 2009), tomato (Purkayastha *et al.*, 2011), garlic slices (Madamba *et al.*, 1996). These mathematical models have been proposed to describe drying characteristics. However, there are still limited studies that have been done on the drying of butterfly pea flowers. This study is aimed at determining the effect of different air temperatures on the drying time of butterfly pea flowers, fitting the drying curves with proposed mathematical models and testing the goodness of fit. Calculation of effective diffusivity and activation energy for the dried butterfly pea flower and analysing their properties after the drying process was done.

2. Materials and methods

2.1 Raw material

Butterfly pea flowers were grown in Vietnam. After manual pickling, approximately 100 to 200 g of the butterfly pea flower was subjected to drying at different temperatures.

2.2 Drying procedure

Four different temperatures (55, 60, 65 and 70°C) were applied and the oven dryer (Model SIBATA SD-60, Japan) was operated at an air velocity of 1.0 m/s. The change of sample weight was recorded by a digital balance (Ohaus, SR series, America, $d = 0.001$) at a half-hourly interval during the drying process. The sample was dried until equilibrium was reached.

2.3 Determination of total anthocyanin content (TAC)

The TAC was determined by the pH differential method, its content was expressed as cyanidin-3-glucoside equivalents as in the following equation (Maran *et al.*, 2015) (Equation 1).

$$\text{Anthocyanin content (mg/L)} = \frac{A \times MW \times DF \times V \times 1000}{a \times l \times m} \quad (1)$$

Where A is the absorbance, MW is the molecular weight of cyanidin-3-glucoside (449.2 g/mol), DF is the dilution factor, V is the solvent volume (mL), a is the molar absorptivity (26,900 L.mol⁻¹.cm⁻¹), and l is the cell path length (1 cm).

2.4 Determination of total phenolic compound (TPC)

The TPC was determined based on the method by Wong *et al.* (2006) with total content was quantified using the standard curve of gallic acid as a standard phenolic compound (0.2 to 1 mg/mL), which was dissolved in deionised water and expressed as mg gallic acid equivalent (GAE) per gram plant material.

2.5 Calculation of drying rate

The drying characteristic of butterfly pea flower was examined using the drying curves and the instantaneous drying rate, DR (g water/g dry matter per min) and it was calculated as Equation 2.

$$DR = \frac{(M_{t+dt} - M_t)}{dt} \quad (2)$$

Where M_{t+dt} and M_t are moisture contents (g water/g dry matter) at time (t + dt) and time t, respectively.

2.6 Mathematical modelling

In order to evaluate the characteristics of the drying process, modelling the drying process is very important. In this study, the drying curves obtained from the experiments were fitted into seven mathematical models that are commonly used for describing the thin layer drying behaviour. These were adjusted to the data from the drying process of the butterfly pea flowers, as shown in Table 1.

In order to obtain the best mathematical model, the moisture content data at different temperatures were converted to Moisture Ratio (MR) that presents the dimensionless moisture ratio using Equation 3.

$$MR = \frac{M_t - M_e}{M_o - M_e} \quad (3)$$

Where M is the instantaneous moisture content (kg_{water} kg⁻¹_{dry matter}) of the product, M_o is the initial moisture content of the product and M_e is the equilibrium moisture content. The values of M_e are relatively negligible compared with M and M_o for a long drying time. Thus Equation 3 has been simplified to Equation 4 (Toğrul and Pehlivan, 2004).

$$MR = \frac{M_t}{M_o} \quad (4)$$

Regression analyses for determining the most suitable model for drying thin layer butterfly pea flower with different temperatures was carried out using the conventional statistical calculations namely the correlation coefficient (R^2), chi-square (χ^2) and root mean square error (RMSE). The highest values of R^2 and the lowest values of χ^2 and RMSE represent the best fitness with experimental data and mathematical models (Akpınar *et al.*, 2003). These statistical values can be

Table 1. Mathematical models used to predict the drying of agricultural products

No.	Models	Equation	References
1	Henderson and Pabis	$MR = a.exp(-kt)$	Rosa <i>et al.</i> (2015)
2	Modified Henderson and Pabis	$MR = a.exp(-kt)+b.exp(-gt)+c.exp(-ht)$	Akpinar and Bicer (2008)
3	Logarithmic	$MR = a.exp(-kt)+c$	Akpinar and Bicer (2008)
4	Newton	$MR = exp(-kt)$	Sobukola <i>et al.</i> (2007)
5	Page	$MR = exp(-k(t^n))$	Akpinar and Bicer (2008)
6	Two-term	$MR = a.exp(-kt)+b.exp(-k_0t)$	Sobukola <i>et al.</i> (2007)
7	Two-term exponential	$MR = a.exp(-kt)+(1-a)exp(-kat)$	Sobukola <i>et al.</i> (2007)

Where t is drying time (hrs); a, b, c, g, h, n, k, k₀ are the model constants.

calculated using Equations 5, 6 and 7 (Akpinar, 2010).

$$R^2 = \frac{\sum_{i=1}^N (MR_i - MR_{pre,i}) \cdot \sum_{i=1}^N (MR_i - MR_{exp,i})}{\sqrt{[\sum_{i=1}^N (MR_i - MR_{pre,i})^2] \cdot [\sum_{i=1}^N (MR_i - MR_{exp,i})^2]}} \quad (5)$$

$$\chi^2 = \frac{\sum_{i=1}^N (MR_{exp,i} - MR_{pre,i})^2}{N - z} \quad (6)$$

$$RMSE = \left[\frac{1}{N} \sum_{i=1}^N (MR_{pre,i} - MR_{exp,i})^2 \right]^{1/2} \quad (7)$$

Where MR_{exp,i} is the ith experimentally observed moisture ratio, MR_{pre,i} is the ith predicted moisture ratio, N is the number of observations and z is the number constants.

2.7 Calculation of the effective moisture diffusivity and activation energy

The drying characteristics of various foods are well described by using Fick’s diffusion equation (John *et al.*, 2014). In the case of thin-layer drying, assuming one-dimensional moisture movement, insignificant shrinkage, constant diffusivity, uniform initial moisture distribution and negligible external resistance (Thorat *et al.*, 2012), a form of the equation given by Crank (1979) can be developed (Equation 8).

$$MR = \frac{8}{\pi^2} \sum_{n=0}^{\infty} \frac{1}{(2n+1)^2} \exp\left(- (2n+1)^2 \pi^2 \frac{D_{eff} t}{4L^2}\right) \quad (8)$$

Where D_{eff} is the effective diffusivity (m²/s), t is drying time (s), n is a positive integer and L is the half thickness of the slab (m).

The linear solution of the equation is obtained by assuming that only the first term in the series equation is significant (n = 0) for long drying times. Then Equation 9 is obtained by taking the natural logarithm of both sides (Akgun and Doymaz, 2005).

$$\ln(MR) = \ln\left(\frac{8}{\pi^2}\right) - \left(\frac{\pi^2}{4L^2} D_{eff} t\right) \quad (9)$$

The diffusion coefficients can be determined by plotting experimental drying data in terms of ln(MR) versus drying time (t) because the plot gives a straight line with a slope as Equation 10 (Zarein *et al.*, 2015).

$$\text{Slope} = \frac{\pi^2 D_{eff}}{4L^2} \quad (10)$$

The activation energy is calculated using the

Arrhenius equation (Equation 11) which expresses the dependence of the effective diffusion coefficient on drying air temperature (Sanjuán *et al.*, 2003).

$$D_{eff} = D_0 \exp\left(-\frac{E_a}{RT}\right) \quad (11)$$

Where D₀ is the diffusion coefficient corresponding to infinite temperature (m²/s), E_a is the activation energy (kJ/mol), R is the universal gas constant (8.314 J/mol K) and T is the absolute drying air temperature (°K).

3. Results and discussion

3.1 Drying kinetics of dried butterfly pea flowers

The effect of temperature used for the thin layer drying process of butterfly pea flower was remarkable with moisture ratios that have decreased continuously in increasing temperature and time (Figure 1). The final moisture content of the butterfly pea flower reached 6 to 8% (DW) after 11, 7.5, 6 and 4 hrs, respectively at drying temperatures of 55, 60, 65 and 70°C. The drying rate is higher when the temperature is increased (Figure 2 and Figure 3). As a result, the time taken to reach the final moisture content is reduced. It was also observed that there is no constant rate drying period in the drying of butterfly pea flower, indicating that diffusion is the main physical mechanism governing moisture migration in the flowers. Similar results were obtained in the study of broccoli (Doymaz, 2014) and okra (Wankhade *et al.*, 2013).

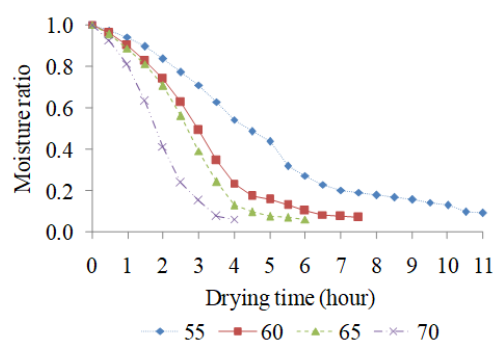


Figure 1. Moisture ratio variation as a function of drying time at different temperatures

3.2 Mathematical modelling

The moisture content data on different experiments were

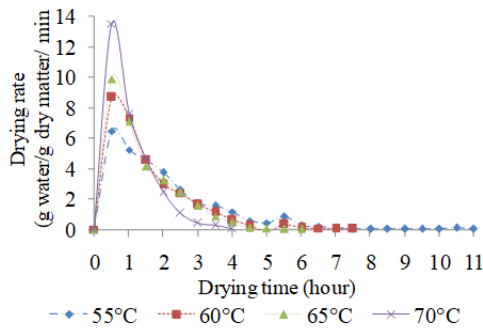


Figure 2. Drying rate of Butterfly pea flower versus drying time

converted to a more useful moisture ratio expression, and curve fitting computations with drying time were performed with the seven drying models. The results of the statistical analyses undertaken on these models for drying are given in Table 2. The models were evaluated based on Root Mean Square Error (RMSE), coefficient of determination (R^2) and Chi-square (χ^2). All equations gave consistently high R^2 values in the range of 0.88 to 0.99. This indicates that all equations could satisfactorily describe the thin layer drying rates of butterfly pea flowers with RMSE ranging from 0.0178 to 0.1658 and Chi-square ranged from 0.0004 to 0.0825.

Among seven thin layer drying models, the Page model obtained the highest R^2 values of 0.9897, 0.9874, 0.9925 and 0.998 at 55°C, 60°C, 65°C and 70°C, respectively. Similarly, the lowest RMSE values were obtained in the Page model over the specified temperature range. Thus, this model is assumed to present the thin layer drying behaviour of the butterfly pea flowers. In a thin layer drying study of persimmon slices, Doymaz (2012) also reported that the suitability of the Page model to fit the experimental drying data of persimmon slices in comparison with other empirical models at 50, 60 and 70°C.

From the results obtained, it could be seen that butterfly pea flowers can best be dried at 55 to 70°C. Comparisons of the predicted values of the Page model with the experimental data are shown in Figure 4 and the Page model displays a good fit to the experimental values. In all temperatures studied, a satisfactory correlation between the predicted MR and the

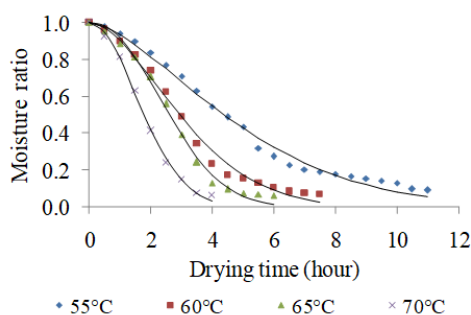


Figure 4. Comparison of experimental data and predicted moisture ratio using Page model for Butterfly pea flowers

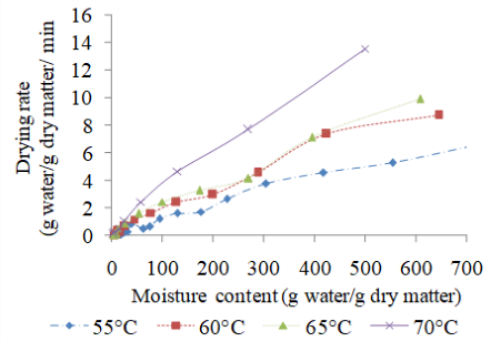


Figure 3. Drying rate of Butterfly pea flower versus moisture content

experimentally determined MR was found with an R^2 value of 0.99 (Figure 5).

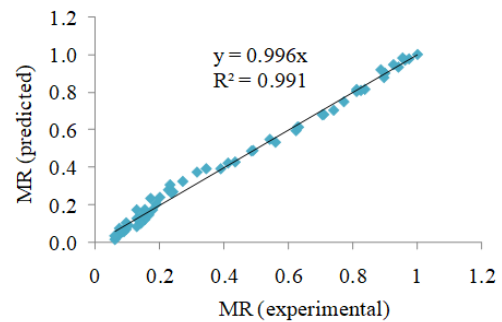


Figure 5. Correlation between the experimentally determined MR values and the MR values estimated for Butterfly pea flower using the Page model

3.3 The moisture diffusivity and activation energy

The effective diffusivity values of dried butterfly pea flowers at 55 to 70°C were varied in the range of 2.392×10^{-12} to 7.756×10^{-12} m²/s. It was observed that the values of D_{eff} increased significantly within creasing temperature. Drying at 70°C gave the highest D_{eff} values. These values are consistent with the estimated D_{eff} values for the *Vernonia amygdalina* leaves (Alara et al., 2019), the effective diffusivities for the three air temperatures ranged (40, 50 and 60°C) from 4.55×10^{-12} to 5.48×10^{-12} m²/s. To obtain the effect of temperature on the effective diffusivity, the values of $\ln(D_{\text{eff}})$ versus $1/T$ (1/K), are plotted as presented in Figure 6. The plot was found to be a straight line over the temperature range investigated, indicating Arrhenius dependency. The activation energy was calculated from the slope of the straight line and was found to be 71.63 kJ/mol.

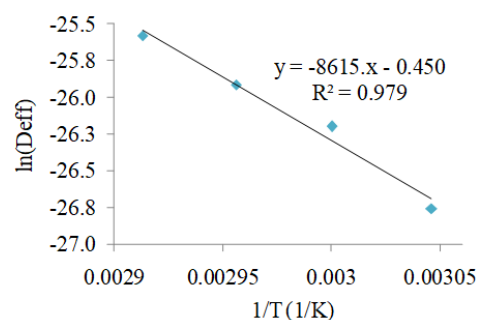


Figure 6. Influence of air temperature on the effective diffusivity

Table 2. Modelling the thin-layer drying process of Butterfly pea flowers at different temperatures

Model	Temperature (°C)	Model constants	RSME	R ² (%)	χ ²
Henderson and Pabis					
	55	a = 1.1385; k = 0.2206	0.0630	96.37	0.0043
	60	a = 1.4162; k = 0.3171	0.0849	94.46	0.0082
	65	a = 1.1589; k = 0.3741	0.1157	91.13	0.0158
	70	a = 1.1253; k = 0.5208	0.1085	92.58	0.0151
Modified Henderson and Pabis					
	55	a = 0.3795; k = 0.2057; b = 0.3795; g = 0.2057; c = 0.3795; h = 0.2057	0.0700	96.37	0.0066
	60	a = 0.3821; k = 0.3172; b = 0.3821; g = 0.3172; c = 0.3821; h = 0.3169	0.1005	94.46	0.0161
	65	a = 0.3863; k = 0.3741; b = 0.3863; g = 0.3741; c = 0.3863; h = 0.3741	0.1450	91.13	0.0391
	70	a = 0.3751; k = 0.5224; b = 0.3751; g = 0.5207; c = 0.3751; b = 0.5194	0.1658	92.58	0.0825
Logarithmic					
	55	a = 1.3600; k = 0.1336; c = -0.2662	0.0520	97.64	0.0031
	60	a = 1.457; k = 0.1830; c = -0.3616	0.0826	95.89	0.0084
	65	a = 2.0367; k = 0.1339; c = -0.9457	0.0826	95.89	0.0179
	70	a = 2.2489; k = 0.1624; c = -1.1825	0.0700	97.36	0.0073
Newton					
	55	k = 0.1791	0.0802	93.84	0.0067
	60	k = 0.2764	0.0996	91.83	0.0106
	65	k = 0.3234	0.1286	88.05	0.0179
	70	k = 0.4626	0.1160	90.31	0.0151
Page					
	55	k = 0.0702; n = 1.5502	0.0320	98.97	0.0011
	60	k = 0.1049; n = 1.749	0.0380	98.74	0.0017
	65	k = 0.0854; n = 2.1826	0.0318	99.25	0.0012
	70	k = 0.2199; n = 1.9744	0.0178	99.80	0.0004
Two-term					
	55	a = 0.5692; k = 0.2057; b = 0.5692; k ₀ = 0.2057	0.0662	96.37	0.0053
	60	a = 0.5731; k = 0.3170; b = 0.5731; k ₀ = 0.3172	0.0917	94.46	0.0112
	65	a = 0.5794; k = 0.3736; b = 0.5794; k ₀ = 0.3745	0.1279	91.13	0.0236
	70	a = 0.5626; k = 0.5211; b = 0.5626; k ₀ = 0.5206	0.1284	92.58	0.0297
Two-term exponential					
	55	a = 1.0342; k = 0.1786	0.0821	93.84	0.0074
	60	a = 1.0189; k = 0.2761	0.1031	91.83	0.0121
	65	a = 0.9870; k = 0.3238	0.1343	88.05	0.0213
	70	a = 1.0031; k = 0.4608	0.1240	90.31	0.0198

The values of activation energy for most agricultural food products lie within the range of 12.7–110 kJ/mol (Akpınar *et al.*, 2003). The E_a obtained from this work is within the general range for food materials, it is found to be lower and higher with other food products reported. This value is higher than that of *Andrographis paniculata* drying 33.4 kJ/mol (Hee and Chong, 2015), okra 51.267 kJ/mol (Doymaz, 2005), Dill and parsley leave 35.05 and 43.92 kJ/mol (Doymaz *et al.*, 2006), Mint leaves 57.12 and 62.96 kJ/mol (Kane *et al.*, 2009;

Doymaz, 2006), but lower than Mint leaves (82.93 kJ/mol) that was reported by Park *et al.* (2002), and black tea 406.028 kJ/mol (Panchariya *et al.*, 2002). Lower activation energy indicates lower sensitivity to air temperature.

3.4 The change of total phenolic and anthocyanin content of butterfly pea flower by drying

The total phenolic (TPC) and anthocyanin content in the butterfly pea flower remained high as drying

temperatures increased, the highest concentration was found at drying temperatures of 65 and 70°C (Table 3). These results may be explained by the long exposure time in the drying process at low temperature (55 to 60°C) and the enzymatic browning reaction that occurred during the drying process. However, the content of TPC and anthocyanin in dried *Clitoria ternatea* were lower than fresh flowers, and after drying, TPC and anthocyanin of samples were ranged from 64.24 to 65.54 mg GAE/g (dwb) and 0.89±0.01 to 1.50±0.04 mg/g flower (dwb), respectively in comparison (as seen in Table 3). Higher drying temperatures (65 to 70°C) not only maintained the bioactive compounds but also improved the colour retention of dried butterfly pea flowers.

Table 3. The bioactive compounds of fresh and dried Butterfly pea flowers

Samples	TPC (mgGAE/g dwb)	Anthocyanin (mg/g dwb)
Fresh Butterfly pea flower		
	67.82±0.13	2.33±0.11
Dried Butterfly pea flower		
55°C	64.24±0.09	0.89±0.01
60°C	64.98±0.11	1.30±0.02
65°C	65.46±0.10	1.47±0.03
70°C	65.54±0.12	1.50±0.04

Values are expressed as the mean±standard deviation. The moisture content of fresh and dried samples were 90.05% and 6-8%, respectively.

4. Conclusion

Thin layer drying characteristics of butterfly pea flowers were investigated at four drying temperatures. As expected, the drying process was shorter at higher drying temperatures. The experimental drying data were fitted to seven empirical mathematical models. Among them, the Page model described the best representation of the experimental drying values at all investigated temperatures (55, 60, 65 and 70°C), with the highest R^2 value and lowest RSME and χ^2 values. Arrhenius' equation was used to evaluate the temperature dependence of the effective diffusivity for calculating activation energy. The findings of this study may be important by providing information for thin-layer drying process conditions of butterfly pea flower from an industrial purpose. The use of temperature from 60 to 70°C for the drying of butterfly pea flower in order to preserve its total phenolic and anthocyanin was also suggested.

Conflict of interest

The authors declare no conflict of interest.

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