Effect of steaming time on antioxidant properties of Napier grass herbal tea by green tea processing method

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Abstract

This study aimed to determine the effect of steaming time on antioxidant properties of Napier grass green tea. Napier grass was subjected to steaming for 1, 2, 3, 4, 6, 8 and 10 mins. Fresh and dried samples were extracted in water (95°C, 30 mins) and the extracts were then analysed by total phenolic content (TPC) assay, total flavonoid content (TFC) assay, diphenylpicrylhydrazyl (DPPH) assay, Ferric reducing antioxidant potential (FRAP) assay, ferric thiocyanate (FTC) method and thiobarbituric acid (TBA) method. Sample steamed for 8 mins showed the highest TPC (18.32±0.26), TFC (152.71±5.74) and 109.88±5.44 in FRAP assay. High antioxidant activity was found in sample steamed for 3 to 10 min (81.63±1.19 to 83.50±1.10) in DPPH which were not significantly different with the fresh sample indicating steaming can retain the phytochemical compounds. Samples undergone 6 to 10 mins steaming time were found to have high lipid peroxidation in ferric thiocyanate (75.02±2.96 to 81.01±6.68) and thiobarbituric acid (85.99±1.56 to 86.21±1.44) assays. The results suggested that 8 mins of steaming time is suitable for Napier grass green tea as it exhibited the greatest antioxidant properties.

1. Introduction

Fruits, herbs, vegetables and various plants are natural sources of antioxidants (Balasundram et al., 2006; Andres-Lacueva et al., 2008; Brito et al., 2014). Numerous previous studies and reviews had reported the benefits of the intake of foods with high contents of natural antioxidants in disease prevention (Duthie and Bellizzi, 1999; Young and Woodside, 2001; Bennett et al., 2012; Hashemi et al., 2014; Sadowska-Bartosz and Bartosz, 2014; Zhang et al., 2015; Rani, 2017). The major compounds that contribute to the antioxidant activity in plants are phenolic compounds such as flavonoids and phenolic acids (Lin et al., 2016).

Various plants are considered as resources of traditional remedies, foods and beverages in Asia Pacific. Malaysia is one of the countries of the Asia-Pacific region which is rich in plant species. Although many plant species have been used for food, industrial, cultural and medicinal purposes, there are still many underutilised plants (Arora, 2014). Napier grass (Pennisetum purpureum) is one of the underutilised plants. In Malaysia, it is commonly used as silage for animal feeds due to its high yield potential and drought tolerance (Lowe et al., 2003; Nyambati et al., 2011). Studies showed that high flavonoids content is found in Napier grass and it is a potential source of minerals, vitamins and dietary protein (Akah and Ani, 2014; Akah and Onweluzo, 2014; Ukpabi et al., 2015). Similarly, tea which is a common drink all over the world is also rich in various nutrients such as antioxidants, amino acids and vitamins (Hui et al., 2004; Mondal, 2007). Although there are studies reported on the high nutrient contents of Napier grass, the usage of Napier grass is restricted by limited knowledge regarding the processing technique in food production. Therefore, it is interesting to explore the potential of Napier grass in the production of herbal tea by tea processing method.

In tea processing, steaming helps to retain the antioxidant components, appearance and texture of the products. It is also one of the important steps in tea making process because over steaming can damage the leaves while insufficient steaming can initiate the onset of the fermentation (Singh et al., 2014). When tea fermentation occurs, organoleptic and antioxidant properties of the tea could be affected (Rodrigues et al., 2014).
2. Materials and methods

Fresh Napier grass was obtained from Kuala Berang, Terengganu. The leaves were cleaned with tap water to clean the soil and dust. After that, cleaned leaves were blotted with tissue papers to absorb water and allowed to dry for 10 mins at 25±1°C.

2.1 Preparation of Napier grass tea by green tea processing method with different steaming time

A total of 50 g of the cut Napier grass (0.5 cm) was steamed for different time (1, 2, 3, 4, 6, 8 and 10 mins) using a steamer pot with 30 cm diameter. The sample was then placed into a 22.9 cm x 16.4 cm zip-lock plastic bag with 0.08 mm thickness and soaked in icy cold water for 30 s to stop the cooking process. Then, the sample was dried at 50°C for 7 h (<6 % moisture content). Dried Napier grass was ground and sieved through a 1 mm metal sieve (Carloni et al., 2013; Lusia Barek et al., 2015 with modifications).

2.2 Preparation of Napier grass green tea infusion (hot water extraction)

Dried Napier grass was ground using a waring blender (Waring Commercial, Torrington. CT, U.S.A) with a particle size less than 1 mm metal sieve. Two grams of powdered Napier grass was infused in 200 mL distilled water (95°C) and stirred for 2 mins using a magnetic stirrer. On the other hand, infusion of fresh Napier grass was used as a control in all experiments. A total of 5 g of fresh Napier grass was ground using a waring blender (Waring Commercial, Torrington. CT, U.S.A). The ground fresh Napier grass was then infused in 200 mL of distilled water (95°C) and continuously stirred for 2 mins using a magnetic stirrer.

All the infusions were left to cool for 30 min of infusion time before filtered through a Whatman No.1 filter paper. After cooling down to room temperature, tea infusions were stored at -20°C until analysis.

2.3 Antioxidant properties assays

2.3.1 Determination of total phenolic content (TPC)

Folin Ciocalteu’s method was used to measure the TPC. Aliquots of sample (1 mL) and gallic acid with different concentrations (10, 20, 40, 60, 80, 100 μg/mL) were transferred into the test tubes. Then, 5 mL of distilled water and 0.5 mL of Folin Ciocalteu’s reagent were added into each test tube and shaken. After 5 mins, 1.5 mL of 20% (m/v) sodium carbonate was added and the volume was made up to 10 mL with distilled water. The samples were incubated for 2 hrs at room temperature. After incubation, absorbance was measured at 750 nm using UV-visible spectrophotometer (SHIMADZU, Kyoto, Japan). The data for TPC of tea infusion was expressed as mg of gallic acid equivalent weight (GAE)/200 mL infusion (Kamtekar et al., 2014).

2.3.2 Determination of total flavonoid content (TFC)

 Aluminium chloride colorimetric assay was used to measure the TFC. A total of 1 mg of aliquots and 1 mL quercetin with different concentrations (100, 200, 400, 600, 800, 1000 μg/mL) were added into test tubes. Then, 4 mL of distilled water and 0.3 mL of 5% (m/v) sodium nitrite solution were added into each test tubes. After 5 mins, 0.3 mL of 10% (m/v) aluminium chloride was added. At 6th min, 2 mL of 1 M sodium hydroxide was added. Finally, volume was made up to 10 mL with distilled water. The absorbance was measured at 510 nm using UV-visible spectrophotometer (SHIMADZU, Kyoto, Japan). The data of TFC of tea infusion was expressed as mg of quercetin equivalents (QE)/200 mL infusion (Kamtekar et al., 2014).

2.3.3 Free radical scavenging ability of DPPH radical (1,1-diphenyl-2-picryl hydrazyl)

DPPH test was performed using the method of Barku, Opoku-Boahen, Owusu-Ansah and Mensah (2013) with some modifications. An aliquot of 2 mL of 0.004% (m/v) DPPH solution in methanol and 1 mL of plant extract were incubated at 25°C for 45 mins. The absorbance of the test mixture was read at 517 nm using a UV-visible spectrophotometer (SHIMADZU, Kyoto, Japan) against a DPPH control containing only 1 mL of distilled water in place of the extract. Percent inhibition was calculated using the following expression:

\[
\text{% Inhibition} = \left( \frac{A_0 - A_t}{A_0} \right) \times 100
\]
Where $A_0$ and $A_t$ is the absorption of the blank sample and the absorption of tested extract solution, respectively.

2.3.4 Ferric reducing antioxidant potential (FRAP) assay

The FRAP reagent was prepared by 300 mM sodium acetate buffer (pH 3.6), 20 mM iron chloride and 10 mM 2,4,6-tripyridyl-s-triazine dissolved in 40 mM hydrochloric acid at a ratio of 10:1:1 (v:v:v). After mixing, the reagent was allowed to incubate at 37°C for 5 mins before use. The initial reading of the reagent was measured at 593 nm using a UV-visible spectrophotometer (SHIMADZU, Kyoto, Japan). An aliquot of 0.1 mL of tea infusion was then added to 2.9 mL of FRAP reagent and kept in the dark for 30 mins. Trolox solution was used to create the calibration curves. Results were expressed as μmol Trolox equivalents (TE)/200 mL infusion (Benzie and Strain, 1996; Heong et al., 2011).

2.3.5 Ferric thiocyanate (FTC) assay

The inhibitory effect of the plant against oxidation by peroxides was evaluated by the modified method of Udaya Prakash et al. (2014). A total of 2 mL of 2.51% (v/v) linoleic acid in ethanol, 120 μL of 98% ethanol and 9 mL of 40 mM phosphate buffer (pH 7) were added to 100 μL of the plant extract. The mixture was incubated in dark, at 40°C. To 100 μL of the mixture, 9.7 mL of 75% (v/v) ethanol, 100 μL of 30% (m/v) ammonium thiocyanate and 100 μL of 20 mM FeCl3 in 3.5% (v/v) HCl were added. The absorbance of the solution was measured at 500 nm, after 3 mins. Butylated hydroxytoluene (BHT) (200 ppm) and α-tocopherol (200 ppm) were used as standards. The percentage of inhibition was calculated using the following equation:

\[ % \text{Inhibition} = \left( \frac{A_0 - A_t}{A_0} \right) \times 100 \]

Where $A_0$ and $A_t$ is the absorption of the blank sample and the absorption of tested extract solution, respectively.

2.3.6 Thiobarbituric acid (TBA) assay

Thiobarbituric acid assay was evaluated following the modified method of Udaya Prakash et al. (2014). A total of 2 mL each of 20% (m/v) trichloroacetic acid (TCA) and 0.67% (m/v) thiobarbituric acid (TBA) were mixed with 1 mL sample. The solution was heated in boiling water bath for 10 mins. After cooling, the solution was centrifuged at 3000 rpm. The absorbance of the supernatant was measured at 532 nm. Butylated hydroxytoluene (BHT) (200 ppm) and α-tocopherol (200 ppm) were used as standards. The percentage of inhibition was calculated using the following equation:

\[ % \text{Inhibition} = \left( \frac{A_0 - A_t}{A_0} \right) \times 100 \]

Where $A_0$ and $A_t$ is the absorption of the blank sample and the absorption of tested extract solution, respectively.

2.4 Statistical analysis

Data were expressed as means ± standard deviation (SD) of triplicates. All data were submitted to one-way analysis of variance (ANOVA) using SPSS (version 20) software. The values were considered to be significantly different when $P<0.05$. The correlations among antioxidant assays were calculated using the Pearson’s correlation coefficient.

3. Results and discussion

3.1 Determination of total phenolic content (TPC)

These findings (Figure 1) showed that the TPC of Napier grass was significantly affected by the steaming time. TPC of Napier grass without steaming (0 min) was 15.37±0.11. The TPC of Napier grass was significantly decreased to 14.24±0.21 at 1 min of steaming time. This might be caused by insufficient steaming of Napier grass that might elevate the degradation of the bioactive compounds (Kessy et al., 2016).

![Figure 1. TPC of infusions of Napier grass undergone different steaming time. Different alphabets on each histogram indicates significant difference at p>0.05.](chart.png)

TPC of Napier grass steamed for 3, 4, 6 and 10 mins were not significantly different with that of fresh Napier grass (16.71±0.15). Moreover, the highest TPC was obtained at 8 mins (18.32±0.26) indicates that steaming can retain and increase TPC in Napier grass. Numerous previous studies reported that steaming can improve antioxidant properties of plant materials. A study by Chumyam et al. (2013) reported that eggplant steamed for 5 to 15 mins was found to have significantly higher TPC than those of raw material. This result was also in accordance with the study by Kessy et al. (2016) which
revealed that litchi pericarp that undergone 3 mins steaming and drying at 60°C showed a higher TPC compared to raw sample. A study of phytochemicals in green leafy Thai vegetables by Nartnampong et al. (2016) also found that asiatic pennywort, garden parsley, peninsular, swamp morning glory and sweet basil showed higher TPC after steaming for 5 mins. Other plant samples such as banana blossom, cauliflower floret, beetroot, tealise gourd, cabbage and tomato also showed an increased TPC on steaming (Saikia and Mahanta, 2013).

Increased in TPC in sample undergone 8 min steaming time might be caused by the breakdown of cellular constituents which release bound phenolic compounds from the ester bonds due to thermal process (Dewanto et al., 2002; Adefegha and Oboh, 2011; Hwang et al., 2012; Saikia and Mahanta, 2013; Kessy et al., 2016; Geetha et al., 2018). The improved release of phenolic compounds might also be caused by the inactivation of oxidative enzymes (Saikia and Mahanta, 2013; Yuniartini et al., 2015; Kessy et al., 2016). In addition, heat treatment causes the degradation of tannins into simple phenolic compounds, whereas steaming treatment leads to the hydrolysis of tannic acid into galloyl like gallotanin, which in turn, increase the TPC (Yuniartini et al., 2015).

In contrast, a study by Hwang et al. (2012) carried out for red pepper demonstrated that steaming for 5 to 15 mins leads to significant loss of TPC in red pepper compared to raw sample. Kalkan and Yucecan (2013) reported that steaming causes loss of TPC in squash, eggplant and potato. Similarly, TPC of green pea, black eye pea, bottle gourd and radish was decreased after steaming for 5 to 6 mins (Saikia and Mahanta, 2013).

TPC in Napier grass was slightly decreased at 10 min steaming time which was 16.23±0.17 compared to the sample steamed for 8 mins which were 18.32±0.26. Factors that cause loss of TPC include processing time and food size (Hwang et al., 2012). Therefore, 10 mins of steaming might start to cause loss of phenolic compound due to thermal degradation of bioactive compounds. A study by Bamidele et al. (2017) reported that further blanching time which more than 5 mins could cause significant reduction in TPC of green leafy vegetables.

3.2 Determination of total flavonoid content (TFC)

Figure 2 shows that TFC of Napier grass undergone steaming process was significantly increased with increasing steaming time until 8 mins (152.71±5.74) compared to the fresh Napier grass and was reduced slightly but not significantly at 10 min (130.93±11.02) compared to the sample steamed for 8 mins. Similar to the result of TPC, it showed that steaming can retain and improve TFC of Napier grass. It also indicated a possible release of certain flavonoids during the steaming of the samples (Kessy et al., 2016). These findings are in accordance with the study by Kessy et al. (2016) which reported that steaming could improve TFC of litchi pericarp. Enhancement of TFC was also reported in green leafy vegetables such as cabbage (Saikia and Mahanta, 2013), Cleome gynandra leaves (Moyo et al., 2016), waterleaf, green amaranth, Chaya and wild basil (Adefegha and Oboh, 2011) after steaming.

TFC of dried Napier grass without steaming (97.82±13.00) was higher than that of fresh (81.38±3.85) and steamed samples have undergone steaming for 1 to 3 mins (80.93±12.22 to 88.93±2.31). TFC of Napier grass might be affected by the preparation process. The dried Napier grass without steaming was dried within the shortest time possible after washing and cutting whereas the steamed samples have undergone 1, 2 and 3 mins steaming showed no significant difference with that of fresh Napier grass indicating 1 to 3 mins steaming was insufficient to inhibit the enzymatic degradation. The short period of steaming might also elevate the degradation of the bioactive compounds because short steaming time could provoke enzymatic reaction (Kessy et al., 2016).

The highest TFC was found in Napier grass steamed for 8 mins (152.71±5.75). There was an agreement between the result of TPC and TFC which found that sample undergone 8 mins of steaming exhibited the highest TPC. Furthermore, this result coincided with the previous study where correlations were established between TPC and TFC (Adefegha and Oboh, 2011).

Numerous previous studies reported that steaming affects the flavonoid content of various plant materials. Based on these previous reports, steaming had positive and negative effects on TFC of plants (Dewanto et al., 2002; Adefegha and Oboh, 2011; Hwang et al., 2012; Saikia and Mahanta, 2013; Kessy et al., 2016; Geetha et al., 2018).
Saikia and Mahanta (2013) revealed that TFC in beetroot, cabbage, radish, tomato, kharua brinjal, knol-knel and carrot were enhanced after steaming. Improvement in TFC possibly because the bound phenolic compounds were released from plant cell constituent that had been thermally damaged during heating (Dewanto et al., 2002; Kessy et al. 2016). In contrast, the negative effects of steaming on TFC was found in banana blossom, cauliflower, green pea, teasle gourd, black eye pea, bottle gourd and roselle leaves (Saikia and Mahanta, 2013).

The presence of phenolic and flavonoids compounds in Napier grass has shown its nutritional value and also its antioxidant role in protecting organism from free radicals that cause ageing and chronic diseases such as cancers, diabetes and cardiovascular diseases (Gramza et al., 2005; Irina and Mohamed, 2012; Kalkan and Yücecan, 2013; Gunathilake and Ranaweera, 2016; Nobosse et al., 2017). Therefore, based on the result of TPC and TFC of this study, consumption of Napier grass tea could be beneficial to body health.

3.3 Free radical scavenging ability of DPPH radical (1,1-diphenyl-2-picryl hydrazyl)

Figure 3 shows that DPPH scavenging activity of Napier grass was significantly changed after steaming. Dried Napier grass without steaming (0 min) showed the highest DPPH scavenging activity (90.02±1.11) followed by samples have undergone steaming for 6 mins (86.14±0.35) and 8 min (86.32±0.09). There is no significant difference between raw sample and samples steamed for 3 to 10 min (81.63±1.19 to 83.50±1.10) indicating that steaming can retain the phytochemical compounds. These phytochemical compounds act as radical scavengers which play an important role in preventing the chain initiation step of reactive species such as free radicals (Adefegha and Oboh, 2011).

Adefegha and Oboh (2011) revealed that DPPH scavenging ability of green leafy vegetables (waterleaf, Senecio biafrae, green amaranth, wild basil, sweet potato leaf, fluted pumpkin and Chaya leaves) was significantly increased after steaming for 10 mins.

In contrast, a study by Preti et al. (2017) carried out for green bean varieties demonstrated that steaming green beans for 15 min lead to loss of DPPH free radical scavenging ability due to thermal degradation. Hwang et al. (2012) also reported that DPPH scavenging activity of red pepper was reduced by 23.5-30.3% after steaming. From the previous study, we can see that steaming could have different effects on the antioxidant activity of different plant materials. Besides cooking time, other factors that affect antioxidant activity of food include method of cooking, temperature and portion size (Hwang et al., 2012).

Although DPPH is a common method used to determine the free radical scavenging ability, it has a limitation of colour interference and sample solubility (Adefegha and Oboh, 2011). Moreover, the complex nature of phytochemicals in plants causes difficulty in evaluating their total antioxidant properties (Gunathilake and Ranaweera, 2016). Therefore, more than one methods such as FRAP, FTC and TBA assays were employed in order to study the antioxidant properties of Napier grass.

3.4 Ferric reducing antioxidant potential (FRAP) assay

FRAP assay was done to study the reducing ability of antioxidants against oxidative effects of reactive oxygen species (Sarla et al., 2011). The findings (Figure 4) showed that antioxidant activity of Napier grass was significantly increased to 91.71±4.14 at 4 mins of steaming time indicating that Napier grass showed increased ferric reducing antioxidant potential upon steaming. Similarly, TFC of Napier grass was also

Figure 3. Percentage of inhibition of DPPH free radical scavenging activity of infusions of Napier grass undergone different steaming time. Different alphabets on each histogram indicates significant difference at p>0.05.

Figure 4. Ferric reducing antioxidant potential of infusions of Napier grass undergone different steaming time. Different alphabets on each histogram indicates significant difference at p>0.05.
increased significantly at 4 mins of steaming compared to the fresh Napier grass.

No significant change was observed between the fresh sample (71.31±1.30) and samples undergone 0 to 3 min of steaming (66.62±2.77 to 76.26±4.39). However, the antioxidant activity of samples undergone steaming (1 to 3 mins) was slightly higher than that of dried sample without steaming (66.62±2.77). This can be explained by the fact that oxidative enzymatic reaction that occurs in the leaves during drying process is terminated by the steaming treatment thereby reducing further oxidative damage during drying (Nobosse et al., 2017).

The highest antioxidant activity was 109.88±5.44 which found in the sample undergone 8 mins steaming. This result is in accordance with that of TPC and TFC indicating that phenolic compounds and flavonoids contribute to the antioxidant activity of Napier grass. Studies show that phenolic compounds in plants act as antioxidants by scavenging free radicals, chelating metals and inhibiting the activity of oxidases (Wiczkowski and Piskula, 2004; Nartnampong et al., 2016).

3.5 Ferric thiocyanate (FTC) assay

FTC assay was done to determine the lipid peroxidation inhibition of Napier grass infusion. The results (Figure 5) show that sample steamed for 10 mins (81.01±6.68) showed the highest percentage of inhibition among the Napier grass samples. This result was different from that of FRAP where 8 mins of steaming was the optimum steaming time for Napier grass. These findings showed that longer steaming time improves antioxidant activity of Napier grass. This might be caused by the termination of enzymatic reaction and the release of certain phenolic compounds from the damaged plant cell constituents during steaming (Dewanto et al., 2002; Adefegha and Oboh, 2011; Hwang et al., 2012; Kessy et al., 2016; Geetha et al., 2018). Phenolic compounds and flavonoids contribute greatly to the antioxidant activity and play role in the prevention of lipids peroxidation (Nobosse et al., 2017).

In FTC assay, lipophilic antioxidants which are α-tocopherol (85.17±2.52) and BHT (85.52±0.30) showed stronger antioxidant activity in emulsions as they concentrate at the lipid/air surface (Kulišić et al., 2006). However, there is no significant difference between the lipid peroxidation inhibition of samples steamed for 6 to 10 mins (75.02±2.96 to 81.01±6.68) and the lipophilic antioxidants used in this assay. This can be explained by the characteristics of plant flavonoids as they can be classified into lipophilic and hydrophilic ones depending on their solubility pattern (Wiczkowski and Piskula, 2004; Saikia and Mahanta, 2013).

3.6 Thiobarbituric acid (TBA) assay

Figure 6 shows the lipid peroxidation inhibition of samples against the secondary stage of lipid peroxidation. Similar to the result of FTC, all the Napier grass samples have lower inhibitory effects than that of positive controls (α-tocopherol and BHT). However, the findings of samples steamed for 6 to 10 mins (85.99±1.56 to 86.21±1.44) have no significant difference with that of positive controls. This showed that the addition of Napier grass infusion to the linoleic acid emulsion was able to reduce lipid peroxidation.

Samples steamed for 1 to 4 mins (72.58±2.94 to 74.63±0.77) showed significant low lipid peroxidation inhibition. However, these samples had no significant difference with that of fresh sample (78.98±10.14) and the sample without steaming (78.16±0.96) indicating 1 to 4 mins of steaming time did not affect the antioxidant activity of Napier grass.

Figure 5. Lipid peroxidation inhibition of infusions of Napier grass undergone different steaming time by FTC assay. Different alphabets on each histogram indicates significant difference at p>0.05.

Figure 6. Lipid peroxidation inhibition of infusions of Napier grass undergone different steaming time by TBA assay. Different alphabets on each histogram indicates significant difference at p>0.05.
Flavonoid aglycones are characterised by hydrophobic properties (Wiczkowski and Piskula, 2004). Most of the flavonoids occur naturally in the glycosidic form except for catechins (flavan-3-ols) which appear as aglycones (Wiczkowski and Piskula, 2004; Thilakaratna and Rupasinghe, 2013). The presence of glycoside in flavonoid molecule increases the hydrophilicity of flavonoids (Wiczkowski and Piskula, 2004). The antioxidant properties of glycosides are usually weaker than those of aglycones, but the bioavailability is occasionally enhanced by glucose (Londoño-Londoño et al., 2010). Therefore, there might be higher content of flavonoids in aglycone form present in the samples undergone steaming for 6 to 10 mins compared to other samples.

4. Conclusion

Steaming time showed a significant effect on the total phenolic and flavonoid contents and antioxidant activity of Napier grass based on DPPH, FRAP, FTC and TBA assays. Napier grass steamed for 8 mins showed the highest value of phenolic and flavonoid content in both analyses. It also showed high antioxidant activity in DPPH, FRAP, FTC and TBA assays. Therefore, 8 mins of steaming time is suitable for the preparation of Napier grass herbal green tea. The dissimilarity of the results might be due to the different antioxidant mechanisms of the methods used or variations of the sample ability to scavenge free radicals.

Conflict of Interest

Authors declare no conflict of interest.

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References


