

Kinetic study, optimization and comparison of sun drying and superheated steam drying of asam gelugor (*Garcinia cambogia*)

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Abstract

The purposes of present study are to compare the kinetic drying of the *G. cambogia* through sun drying and superheated steam drying (SSD) method and optimizing the quality of SSD of it through response surface methodology. *G. cambogia* fruit rinds were dried at temperature of 150°C, 200°C and 250°C. The drying curves were fitted into the mathematical model of Page, Lewis and Henderson-Pabis models. Page model was found to be the best in describing the drying behavior of *G. cambogia*. Drying rate constant (k) increased as temperature increased and SSD method had overall higher drying rates ranged from 5.929×10^{-5} to $5.861 \times 10^{-4} \text{ min}^{-1}$ than sun drying method which was $4.980 \times 10^{-6} \text{ min}^{-1}$. Total acid number showed a trend of increased followed by decreased over drying time. superheated steam drying process of *G. cambogia* fruit rinds was optimized by using response surface methodology employing a central composite design. Drying time and temperature were the factors in optimization while moisture content (wet basis), acid number and lightness (*L) were the response parameters. Experimental results were fitted to a second-order polynomial model and the model fitness and optimal drying condition were determined by regression analysis and analysis of variance. The optimal conditions for superheated steam drying of *G. cambogia* fruit rinds were identified as 46.60 mins and 150°C with the composite desirability of 0.913. Application of superheated steam drying under controlled conditions resulted in faster drying process and better quality of dried *G. cambogia* than conventional sun drying technique.

1. Introduction

Garcinia is the largest genus of Guttiferae (syn. *Chusiaceae*) family. It is a native species distributed in India, Nepal and Sri Lanka and has been brought to China, Malaysia and Philippines (Jayaprakasha, 2002). It produces small, green fruit about 1.5 inches in diameter that resemble pumpkins and changes from green to yellow colour during the ripening process. The fruits are edible however too acidic. Various parts of the fruit are used as food preservatives, spices, and food bulking agents. It is most often used for culinary especially fish curries and 'Asam laksa' because of the sour taste and medicinal purposes (Osman and Milan, 2006). The fruit extract of *G. cambogia* has been used for constipation, hemorrhoids, and intestinal parasites. Other medicinal uses include edema, rheumatism, and irregular menstruation. Although numerous chemicals have been isolated from *G. cambogia* fruit, hydroxycitric acid (HCA) is considered the active ingredient for weight loss (Haber *et al.*, 2018). Several clinical studies on the

efficacy of *G. cambogia* for obesity have been conducted (Upadhyay *et al.*, 2013; Gogoi *et al.*, 2014). Due to its excellent therapeutic value, *G. cambogia* are being planted and produced in mass quantities and is an important economic product in India.

Thermal processing is one of the most important methods of food preservation primarily intended to inactivate enzymes, deteriorative microorganisms and reduce water activity by dehydration. For decades, conventional sun drying (SD) method was applied to enhance the shelf life of the fruits due to low drying cost. The traditional method of drying is by slicing *G. cambogia* into thin slices and dehydrated under the sun. However, the quality of the dried fruits is difficult to control as drying proceeds in an open environment (Rittirut and Siripatana, 2006)

Superheated steam is generated when steam is further heated to raise the temperature above saturation temperature. The properties of superheated steam and the benefits it brings gain more acceptance and has been

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used to dehydrate a variety of products. According to Mujumdar (2006), superheated steam drying (SSD) can better preserve dried food quality due to lack of air. Recent application of superheated steam in food products are banana slices (Hamawand *et al.*, 2014), potato chips (Caixeta *et al.*, 2002), shrimps (Prachayawarakorn *et al.*, 2002), instant Asian noodles (Pronyk *et al.*, 2004), vegetables, herbs and spices (Van Deventer and Heijmans, 2001).

To the best of our knowledge, no research has been carried out on the drying of *G. cambogia* by superheated steam drying (SSD) method and compared it with sun drying (SD). Hence, current study was conducted to improve the dried quality of *G. cambogia* by application of superheated steam in the drying process under controlled conditions. The kinetic drying of superheated steam drying method and sun drying method were determined and optimization of the superheated steam dried quality of *G. cambogia* was done by using response surface methodology (RSM).

2. Materials and methods

2.1 Materials

2.1.1 *Garcinia cambogia* fruit

Ripen *G. cambogia* fruits are freshly harvested on a weekly basis from an orchard located in Selama, Perak, Malaysia. The fresh fruits are wrapped with papers, kept in plastic bags and transported to the laboratory. The fruits are stored in chiller (Panasonic, NR-BL307N, Indonesia) at $10 \pm 1^\circ\text{C}$ to maintain its freshness for further use. Each *G. cambogia* fruit weighs 300 to 350 g and depending on the fruits' sizes, two to three fruits are needed for each superheated steam drying in order to obtain kinetic drying curve.

2.1.2 Chemicals and reagents

Ethanol (approximately 96%), 2-propanol and toluene were purchased from QR $\text{\textcircled{R}}$ CTM, New Zealand. Potassium hydroxide was obtained from System $\text{\textcircled{R}}$, Shah Alam, Malaysia. Phenolphthalein was purchased from Sigma-Aldrich, St. Louis, Missouri, United States. All the chemicals and reagents purchased were analytical grade.

2.2 Sample preparation

G. cambogia fruits were washed thoroughly to remove dirt and gums on the surface. The fruits were then patted dry with a paper towel and sliced into thin slices approximately 2.0 ± 0.5 mm without removing the skin to prevent moisture evaporate and diffuse off from the thin slice in radial dimension. Sliced fruits were kept in the resealable bag prior to drying.

2.3 Drying kinetics of superheated steam drying and sun drying of *G. cambogia*

2.3.1 Superheated steam drying

Sliced *G. cambogia* fruits were spread in a single layer on drying trays. Superheated steam oven (SHARP Healsio, AX-1500 (K), Osaka, Japan) was turned on to the superheated steam mode and was preheated to the desired temperature. Superheated steam drying of *G. cambogia* was done at 150, 200 and 250°C . During the drying process, samples were taken at different time intervals and instantaneously cooled to room temperature in dessicator. Samplings were done at 0, 10, 20, 30, 40, 50, 60, 75 and 90 mins. Collected samples were blended using a blender (PENSONIC, PB-3205, Malaysia) and packed in resealable polyethylene plastic bags and stored for further experiments.

2.3.2 Sun drying

A single layer sliced *G. cambogia* fruits were distributed on a drying mesh and dried under the sun. The fruit slices were left exposed to the sun in an open environment to imitate the commercial sun drying method. However due to time constraints, sun drying of samples was done for 6 hrs only. Samplings were done at time intervals: 0.0, 0.5, 1.0, 1.5, 2.0, 2.5, 3.0, 4.0, 5.0 and 6.0 hrs. The collected samples were blend, packed and stored for further analysis similar to superheated steam drying samples.

2.4 Thin layer model

Page model, Lewis model, and Henderson-Pabis model were used to describe the drying behavior of *G. cambogia* fruit rinds. These three semi-theoretical models have been widely used to quantify drying kinetics of various foodstuffs such as mushrooms (Giri and Prasad, 2007), basil leaves (Kadam *et al.*, 2011) and lactose powder (McMinn, 2006). The models were derived from Fick's law by simplification and are less time consuming than theoretical models in calculations (Guillard *et al.*, 2013). Equation 1 shows the equation suggested by Lewis model.

Page (Page, 1949) then proposed a model to overcome the shortcoming of Lewis model by introducing an empirical parameter, η resulting in Equation (2):

$$MR = \frac{M - M_e}{M_i - M_e} = \exp(-kt) \quad (1)$$

$$MR = \frac{M - M_e}{M_i - M_e} = \exp(-kt^n) \quad (2)$$

Henderson and Pabis (Henderson, 1974) did some modification on Lewis equation, by adding a constant a

in the exponential term (Equation (3)).

$$MR = \frac{M - M_e}{M_i - M_e} = a \exp(-kt) \tag{3}$$

Where MR is the moisture ratio; M is the moisture content at any time t ; M_e is the equilibrium moisture content; M_i is the initial moisture content; k is the drying constant (min^{-1}); n and a are model constants (dimensionless)

The equation proposed in Page model, Lewis model and Henderson-Pabis model (Equations 1-3) were linearized and the empirical constants, a and n and drying rate constant, k were determined and evaluated according to coefficient of determination, R^2 (Roberts *et al.*, 2008). The linearized Lewis equation is in the form of

$$\ln(MR) = -kt + 1 \tag{4}$$

where the drying constant, k is the gradient of $\ln(MR)$ versus t curve. The form of linearized Page equation is

$$\ln[-\ln(MR)] = \ln(k) + n \ln(t) \tag{5}$$

where the drying constants, k and dimensionless constant, n were determined from the intercept and slope of the $\ln[-\ln(MR)]$ versus $\ln(t)$ curve respectively. The linearized Henderson-Pabis equation is in the form of

$$\ln(MR) = -kt + a \tag{6}$$

where the drying constants, k and dimensionless constant, a were determined from the intercept and slope of the $\ln(MR)$ versus time curve.

The calculated constants were evaluated according to R^2 (coefficient of determination) value which higher R^2 indicated the model is fitted to the kinetic drying curve of *G. cambogia*. RMSE (root mean square error) and X^2 (chi-square) values were also calculated to better support R^2 value before drawing conclusion on the model which best fitted the drying curve (Onwude *et al.*, 2016). The RMSE and X^2 is calculated using Equation (7) and (8), respectively.

$$RMSE = \sqrt{\frac{1}{N} \sum_{i=1}^N (MR_{exp,i} - MR_{pred,i})^2} \tag{7}$$

$$\chi^2 = \frac{\sum_{i=1}^N (MR_{exp,i} - MR_{pred,i})^2}{N - n_p} \tag{8}$$

Where; $MR_{exp,i}$ is the experimental moisture ratio; $MR_{pred,i}$ is the predicted moisture ratio, N is the number of experimental data points and n_p is the number of parameters in model.

2.5 Optimization of superheated steam drying

G. cambogia samples were prepared following the

central composite design (CCD) which each numeric factor is varied over 5 levels that are $\pm \alpha$ (axial points), ± 1 (factorial points) and center point. In this experimental design, α of 1.0 (face centered) option was chosen so that all the parameters were within the desired range. Number of design points generated were 8 ‘not center point’ and 5 ‘center points’, giving a total of 13 assays from two independent numeric factors, time and temperature. The design points of time and temperature were varied, in the range of 45 to 50 mins and between 150 to 250°C. Table 1 shows the generated design points for *G. cambogia* fruit rinds using Minitab software (Version 16.1.1, Minitab Inc., USA). The dependent variables or responses are moisture content, colour parameter, L^* (lightness) and acid number. The experiment was randomized, repeated twice for the entire set of 13 assays and mean values from the experiments were taken for optimization analysis.

Table 1. The experimental design for optimization of *G. cambogia* fruit rinds

Run Order	Codified		Decodified	
	Time	Temperature	Time (min)	Temperature (°C)
1	1	0	55	200
2	0	0	50	200
3	-1	1	45	250
4	0	0	50	200
5	0	1	50	250
6	-1	0	45	200
7	1	1	55	250
8	0	0	50	200
9	1	-1	55	150
10	0	0	50	200
11	0	-1	50	150
12	-1	-1	45	150
13	0	0	50	200

2.6 Moisture content determination

Moisture content analysis (Chemists and Horwitz, 1990) was done on the samples collected in duplicate. Moisture can with lid was cleaned and dried in hot air oven (Memmert, UM600, Schwabach, Germany) for at least 2 hrs at 105°C. The dried moisture can and the lid were removed from hot air oven and left to cool in a desiccator. The weight of the moisture can with lid was measured by digital balance and the reading was recorded. Approximately 2 to 3 g of the blended sample was weighed into the moisture can and dried in hot air oven at 105°C for 16 hrs. The moisture can was kept semi-covered by the lid to allow moisture to leave the sample. After dried for overnight, moisture can was removed from hot air oven, realigned lid to close and cooled in desiccators until the weight of moisture can and the sample was determined. The calculation was performed to obtain the percentage of moisture content on the wet basis of the samples.

2.7 Titratable acidity

Preparation of chemical required for titratable acidity test was done prior to the test. Toluene-isopropyl alcohol mixture was prepared by mixing toluene and 2-propanol in 1:1 ratio. 0.1 N alcoholic KOH was prepared according to procedures listed by Pharmaceutical Guidelines (2015). 6 g of KOH was weighed and dissolved in 5 mL of distilled water. The solution was then transferred to a 1000 mL volumetric flask and top up with 95% ethanol until the benchmark to produce 1000 mL solution. The solution was stoppered and allowed to stand for 24 hrs. The clear solution on the top was transferred in a tightly closed container and standardization of the solution was carried out. Approximately 1 to 2 g of blended sample was weighed into 100 mL beaker. 50 to 100 mL toluene-isopropyl alcohol mixture was added to the sample and left to stand for 10 mins. The mixture was then filtered using Whatman® Grade 1 qualitative filter paper into a conical flask and added with 0.1 ml of phenolphthalein solution. Titration with 0.1 N alcoholic KOH was performed until a permanent faint pink which persists for at least 10 seconds appeared.

Acid value of the sample was calculated as follow:

$$\text{Acid value} = \frac{\text{mL of alc. KOH soln.} \times 0.1 \text{ N} \times 56.1}{\text{Weight of sample (g)}} \quad (9)$$

Note that difference between duplicate should not exceed 0.1 mg KOH/g sample.

2.8 Colour measurement

L^* (lightness) of blended samples were measured by using spectrophotometer (KONICA MINOLTA, CM-3500d, Osaka, Japan). The measurement was performed via SpectraMagic™ NX computer software and calibration was carried out according to the user manual for powder before sample measurement. First, target mask for petri dish (CM-A127) was placed on the instrument's shutter. The instrument setting was changed to reflectance: $d/8$ (diffuse illumination), specular component excluded (SCE), large aperture and target sample holder petri dish. Then, aperture of target mask was covered by zero calibration box (CM-A124) to perform zero calibration followed by white calibration with white calibration plate (CM-A120). After completed the calibration steps, measurement of samples can proceed. The sample was filled into a petri dish and placed on the target mask then covered with zero calibration mask. The measurement on target sample was taken for 3 replications and data were presented in an average.

2.9 Statistical analysis

All data generated were analyzed with Minitab software (Version 16.1.1, Minitab Inc., USA). Analysis of variance and RSM were conducted. All the responses (L^* value of colour, moisture content, and acid value) were analyzed to diagnose the significant effects of each factor over each response. The data of responses in relation to factors was each fitted into a second-order polynomial equation in a function of X .

$$Y = \beta_0 + \beta_1 X_1 + \beta_2 X_2 + \beta_{12} X_1 X_2 + \beta_{11} X_1^2 + \beta_{22} X_2^2 \quad (10)$$

Where; Y is the predicted response; β_0 is a constant coefficient; β_1 and β_2 are linear regression coefficients; β_{12} is the interaction regression coefficient; β_{11} and β_{22} are quadratic regression coefficients; X_1 is the coded value (+1, 0, -1) of factor, time; and X_2 is the coded value (+1, 0, -1) of factor, temperature.

R^2 values of the regressions to above 90% were used to indicate the adequacy of the models and insignificant lack-of-fit test ($p > 0.05$) as good indicators of the fitness. Optimization analysis was carried out next using the same software. The desirability functions proposed by Derringer (1980) were used to determine the optimum point prediction for drying of *G. cambogia*. The desirability function method transforms each response into an individual desirability value by defining the goals and boundaries and later combined all individual desirability into a single measure using a geometric mean (Wan and Birch, 2011). The highest desirability value is 1. The optimum quality of *G. cambogia* was determined following target values of moisture content of 15.50%, with maximum of L^* and acid number. The moisture content of 15.50% refers to the commercially available dried *G. cambogia* products purchased from a local market in Penang, Malaysia.

3. Results and discussions

3.1 Kinetic study

3.1.1 Kinetic drying curves of *Garcinia cambogia* using SSD

G. cambogia fruit rinds with initial moisture content approximately 87% on wet basis were dried in a superheated steam oven until its equilibrium moisture content was reached. The kinetic drying curves of *G. cambogia* fruit rinds at temperatures of 150, 200 and 250°C are shown in Figure 1a. The moisture content of *G. cambogia* fruit rinds decreased with time when heat from the superheated steam withdrew the moisture away from the thin layer. The kinetic drying curve of superheated steam drying at 250°C decreased in the steepest manner followed by 200°C and lastly the least steep at 150°C (Figure 1a). The different sloppiness of

the curves between different temperatures was due to the different drying rates. The kinetic drying of *G. cambogia* fruits demonstrated two distinct drying rates which are constant drying rate and falling rate. In the first 15 mins period, constant drying period occurred which moisture is freely available on the surface and immediate moisture removal was allowed through evaporation. After the surface moisture had been exhausted, the process went into falling rate of drying period which internal diffusion of moisture to the surface was required to remove the moisture. The amount of moisture available became progressively scarce and the drying rate decreased with time until constant equilibrium moisture content was achieved.

Generally, drying at higher temperature increases the drying rate and equilibrium moisture content can be achieved in a shorter time than lower temperature. As shown in Figure 1a, equilibrium moisture content was achieved at 55 mins of drying when drying was performed at 250°C whereas required more than 90 mins

for drying at 150°C. The higher drying temperature implies larger driving force for heat transfer and leads to higher values of moisture diffusivity (Methakhup *et al.*, 2005). The idea of the relationship between temperature and moisture diffusivity can be better known through the temperature dependent effective diffusivity (D_{eff}) equation which followed an Arrhenius relationship (Suarez *et al.*, 1980).

3.1.2 Comparison of *Garcinia cambogia* by drying kinetic of SSD and SD

From the results obtained, the objective of the study was achieved as superheated steam drying was more efficient in term of time saving. The drying time of *G. cambogia* fruit rinds was reduced to an hour time as compared to sun drying under an open environment which was truly affected by the weather, humidity and air velocity which took 6 hrs and only happened to reduce the moisture content to 17.47% on a wet basis (Figure 1b). However, drying of *G. cambogia* fruit rinds

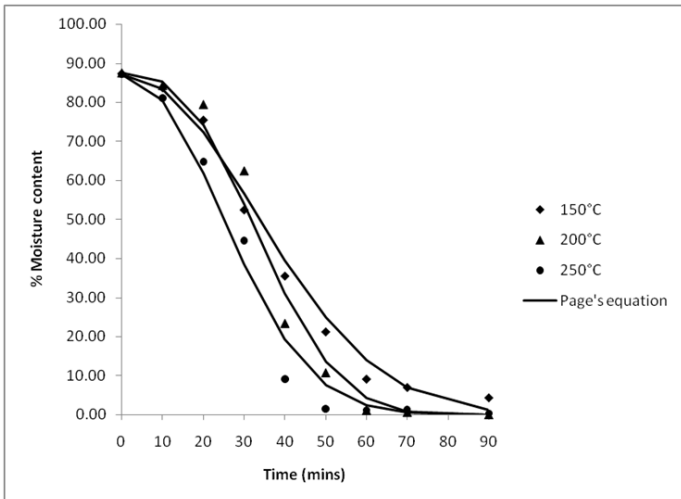


Figure 1a. Drying curves of *G. cambogia* fruit rinds by using superheated steam drying at temperature 150, 200 and 250°C fitted according to Page's equation.

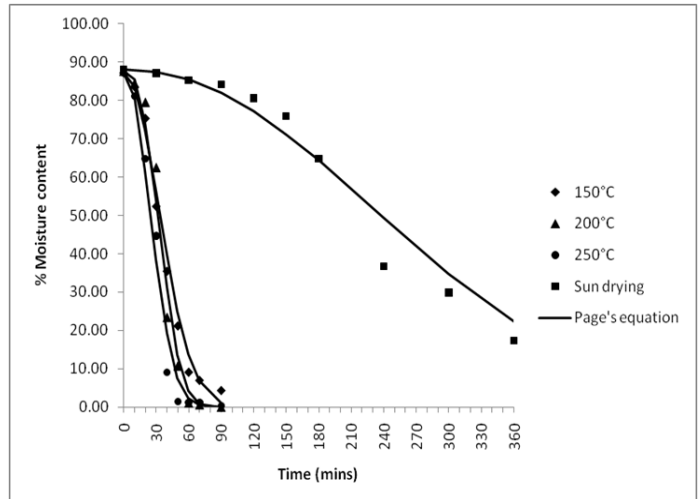


Figure 1b. Comparison of drying kinetic of *G. cambogia* fruit rinds by superheated steam drying and sun drying with data fitted according to Page's equation.

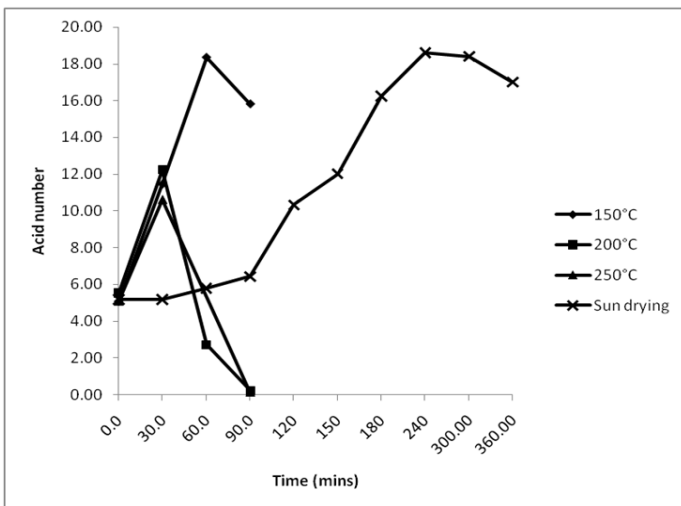


Figure 1c. The relationship of acid number with drying time at 150, 200 and 250°C of superheated steam drying and sun drying.

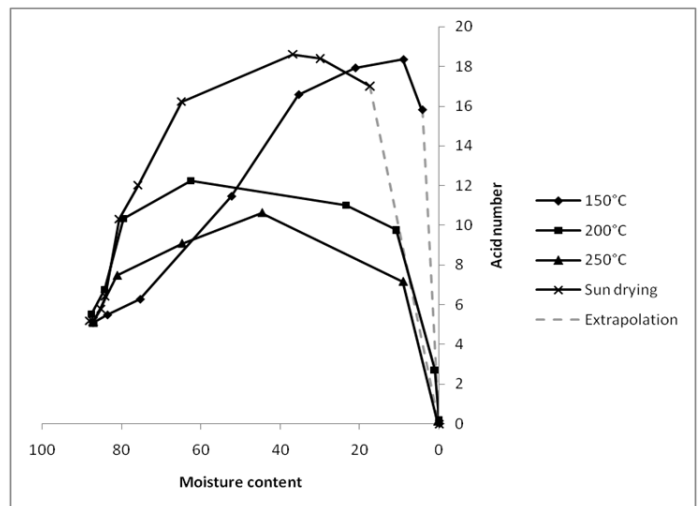


Figure 1d. The relationship of acid number with moisture content during drying process

at high temperature is not favourable even though shorter drying time can be achieved. High heat treatment on the fruit would eventually cause thermal damages to organic substances presence in the fruit and affect the overall final product quality. *G. cambogia* is rich in (-) hydroxycitric acid (HCA) which has antiobesity potency (Gogoi et al., 2014). The organic acids also contribute to the characteristic sourness (Raju and Reni, 2001) and thus often *G. cambogia* dried fruit rinds are used as flavouring and condiments in Indian and Thai cuisine (Great Recipe Tips, 2014). Hence, the consideration of temperature used in superheated steam drying is significant in order to achieve a balance in drying time and quality.

3.1.3 The relationship of acidity and drying time of SSD and SD

The relationship of acid number and drying time at different drying temperatures were illustrated in Figure 1c. The acid number was found to have a common trend which the value increase with drying time until a turning point, decreased when further dried. This trend was clearly seen in *G. cambogia* fruit rinds dried by the superheated steam oven at 150, 200 and 250°C. Superheated steam drying at 200°C and 250°C showed acid numbers both peaked at 30 mins of drying while for drying at 150°C the acid number peaked at 60 mins of drying. For sun drying, a continuous increased of the acid number was observed till the fourth hour of the drying period.

3.1.4 The relationship of acid number with moisture during drying process

Relationship of acid number and moisture content at different drying temperatures were illustrated in Figure 1d. The experimental data suggested that the rise in acid number can be attributed to initial heat provided during the drying process, however, was limited by moisture content and further heating. An illusion line was used to give a better clarity on the effect of reduced in moisture content to the acid number.

The early stage of drying yields a higher acid number which does not correspond with the general perception that phytochemical compounds will be destroyed or damaged by high heat. This outcome opposed the research by Pacheco-Palencia et al. (2009) and Song and Milner (2001) which discuss on thermal stability of phytochemicals in Acai fruit and garlic

respectively. However, research reported by Chan et al. (2008) and Zhu (1998) supported the experimental outcome as research showed that heat treatments on herbs lead to the elimination, separation or release of phytochemical element which in turn changes the biological nature or creation of new biological characteristic in herbs extract. Lactonisation of hydroxycitric acid which usually happened during evaporation and concentration (Jayaprakasha and Sakariah, 2002) is likely to contribute to the high acid number in the early stage of drying. Further heating of fruit rinds affects the thermal stability of organic acids and also with a reduction in moisture content lead to destroy and breakdown thus having the acid number in decreasing trend with drying time.

3.1.5 Modeling the drying behavior of *Garcinia cambogia*

In this study, equilibrium moisture content (M_e) for sun drying method was unable to determine as drying only took place in a duration of 6 hours and the moisture content has not yet to reach equilibrium moisture content. To achieve moisture level of 15 to 20%, sun drying of *G. cambogia* under open environment required three to seven days under thin layer conditions (Raju and Reni, 2001). In order to provide a similar final state of drying, equilibrium moisture contents for all drying conditions were assumed to be 0% in wet basis. Several papers also adapted this assumption on equilibrium moisture content since the value was significantly less than initial moisture content and some with the reasons of fluctuating relative humidity during the drying process (McMinn, 2006). After obtaining data of moisture ratio at each drying time, drying rate constant (k) and dimensionless constants (a and N) were determined from the intercept and slope of normalized equation of each model respectively. Page model, Lewis model and Henderson-Pabis model mathematical and normalized expression were shown as in Table 2. Empirical constant of Page model, Lewis model and Henderson-Pabis model equations were tabulated in Table 3 with coefficient of determination (R^2) indicating the fitness of each model to the drying kinetic of *G. cambogia* fruit rinds. RMSE (root mean squared error) and X^2 (chi-square) were used to check the ability of the model to represent the drying data. Calculation of RMSE and X^2 in Table 4 were based on equation (7) and equation (8) respectively.

Higher values of R^2 suggested that the particular

Table 2. Thin layer models fitted to experiment data

Model	Mathematical expression	Normalized expression
Page	$MR = \exp(-kt^n)$	$\ln[-\ln(MR)] = \ln(k) + n \ln(t)$
Lewis	$MR = \exp(-kt)$	$\ln(MR) = -kt + 1$
Henderson-Pabis	$MR = a \exp(-kt)$	$\ln(MR) = -kt + a$

Table 3. Empirical constant of Page, Lewis and Henderson-Pabis equation

Temperature	Page equation			Lewis equation		Henderson-Pabis equation		
	k (min ⁻¹)	n	R ²	k (min ⁻¹)	R ²	k (min ⁻¹)	a	R ²
150°C	3.612 x 10 ⁻⁴	2.084	0.982	0.048	0.849	0.038	0.377	0.951
200°C	5.929 x 10 ⁻⁵	2.647	0.969	0.08	0.905	0.083	1.198	0.907
250°C	5.861 x 10 ⁻⁴	2.129	0.956	0.077	0.897	0.069	0.518	0.915
Sun drying	4.980 x 10 ⁻⁶	2.128	0.962	0.007	0.253	0.004	0.274	0.885

Table 4. Model prediction evaluation

Temperature	Page equation		Lewis equation		Henderson-Pabis equation	
	RMSE	X ²	RMSE	X ²	RMSE	X ²
150°C	0.0364	1.863 x 10 ⁻⁸	0.6247	3.691 x 10 ⁻¹	0.1634	2.294 x 10 ⁻²
200°C	0.0505	5.022 x 10 ⁻¹⁰	0.6143	3.691 x 10 ⁻¹	0.8084	7.747 x 10 ⁻¹
250°C	0.0527	4.095 x 10 ⁻⁸	0.5913	3.691 x 10 ⁻¹	0.2714	6.580 x 10 ⁻¹
Sun drying	0.0561	3.101 x 10 ⁻¹²	0.7363	3.691 x 10 ⁻¹	0.1412	1.242 x 10 ⁻²

model has a better prediction on the drying behaviour of *G. cambogia* fruit rinds (Giri and Prasad, 2007; Onwude et al., 2016). The reduced values for RMSE and X² also represents the ability of the model to represents the drying data. Reduced X² takes in the number of constants in the model and with the magnitude of this parameter the model is described to have higher reliability in presenting the drying data (McMinn, 2006). By various means of statistical parameters, the best fit was decided for highest R² and minimum RMSE and X² (Kadam et al., 2011). From Table 3 and Table 4, Page model equation was showed as best model fitted to the experimental drying data of superheated steam drying at 150, 200 and 250°C and sun drying. The R² values obtained were relatively higher range from 0.956 to 0.982 compared to Lewis model equation and Henderson-Pabis model equation with significant reduced in RMSE and X² values.

Overall, drying rate constants (k) of *G. cambogia* fruit rinds by using superheated steam was higher as compared to sun drying. Drying rate constant (k) increased as temperature increased and SSD method had overall higher drying rates ranged from 5.929 x 10⁻⁵ to 5.861 x 10⁻⁴ min⁻¹ than sun drying method which was 4.980 x 10⁻⁶ min⁻¹. The higher drying rate indicated that time required for reducing moisture content of *G. cambogia* to a desire level has been shortened. This implies that drying of fruit rinds by using superheated

steam oven is much faster than sun drying. Drying rate also showed linear relationship with superheated steam drying temperature which at 150°C was 3.612 x 10⁻⁴ min⁻¹ and increased to 5.861 x 10⁻⁴ min⁻¹ at 250°C in Page model. Similar trend was observed in Lewis model equation and Henderson-Pabis model equation too which the drying rates increased with temperature increased. The increased in drying rates with increased of drying temperature were due to effective diffusivity which is temperature dependent (Vaccarezza et al., 1974). The increased in drying temperature causes the driving force for heat and mass transfer to increase and hence increase the drying rate (Methakhup et al., 2005).

3.2 Response surface methodology and optimization

Optimization of drying of *G. cambogia* fruit rinds was done by response surface methodology. Regression coefficients of the second-order polynomial model presented in Table 5 yield the following equations:

Y_1 , Y_2 and Y_3 represents moisture content, acid number and lightness, respectively.

The independent and dependent variables were fitted to the equations and examination of the goodness of fit was carried out by determining R² values and analysis of variance on the effect of time and temperature of drying towards the responses as shown in Table 6. The results denoted that the models for moisture content, acid

Table 5. Regression coefficients and the associated probability (P-value) of the second-order polynomial response models

Term	Moisture content, Y ₁		Acid number, Y ₂		Lightness, Y ₃	
	β	P-value	β	P-value	β	P-value
Constant	0.5462	0.38	0.7352	0.067	18.9555	0.000*
X ₁	-2.9133	0.001*	-1.7517	0.001*	-0.2533	0.38
X ₂	-5.385	0.000*	-5.015	0.000*	-2.0367	0.000*
X ₁ X ₂	2.1925	0.017*	1.6975	0.004*	-0.0475	0.89
X ₁ ²	2.5083	0.021*	0.1619	0.752	0.8507	0.070*
X ₂ ²	5.0233	0.001*	4.6019	0.000*	0.6707	0.137

β is the regression coefficient. X₁ and X₂ are coded factor for time and temperature respectively. *p<0.05 denotes that there is significant effects

number and lightness are at satisfactory level as the R^2 are particularly high for all models ($R^2 > 0.900$). Results showed that there was no significant lack of fit ($p > 0.05$) in response of acid number and lightness. However, moisture content was significant lack of fit suggested that the model was not applicable to represent the data. Repetition on drying procedures were performed in order to improve the lack of fit, but was unsuccessful. Pua *et al.* (2007) reported that lack of fit may be due to the large experimental region covered in a study. This report is invalid to explain current lack of fit as only two independent variables involved in this model. While Box and Draper (1987) explained significant lack of fit of a model might due to the lack of variable in the experimental study which appropriate functions of independent variables were not included. After all, the model with a significant lack of fit could still be used if most of the models showed no significant lack of fit. Further verification of the results will be needed to ensure the usage is unlikely to be questionable and values obtained through calculations using model equation shall possessed good agreement with the experimental value (Torreggiani *et al.*, 1995).

Table 6. Analysis of variance on the independent variables as linear, quadratic and interaction terms on the response variables

Source	df	Sequential sum of squares		
		Moisture content	Acid number	Lightness
Regression	5	377.007*	251.175*	30.4779*
Linear	2	224.914*	169.311*	25.2731*
Square	2	132.864*	70.338*	5.1957*
Interaction	1	19.228*	11.526*	0.009
Residual error	7	13.78	14.692	3.0757
Lack of fit	3	12.657*	3.438	1.4603
Pure error	4	1.123	1.254	1.6155
Total	12	390.787	255.867	33.5536
R^2 (%)		96.47	98.17	90.38

* $p < 0.05$ denotes that there is significant effect.

From Table 6, time and temperature were showed to be significant factors ($p < 0.05$) for moisture content, acid number, and lightness. Time and temperature appeared to have significant effects on moisture content and acid number in linear, quadratic and interaction terms whereas only linear and quadratic terms showed the significant effect to lightness. From Table 5, it was observed that significant first and second order polynomial models were found for moisture content which had a negative linear effect and positive non-linear effect with drying time and temperature. Figure 2a was the surface plot of moisture content to temperature and time. The surface plot indicated that moisture content decreased with the increased of temperature and time. The increased in temperature created a larger driving

force for heat and mass transfer and thus promote efficient diffusivity which allowed moisture to leave the drying surface. During drying, moisture content was evaporated and continuously diffused to the surface. As drying time was prolonged, more and more moisture content can be drawn out from *G. cambogia* fruit rinds.

Surface Plot of Moisture Content vs Temperature, Time

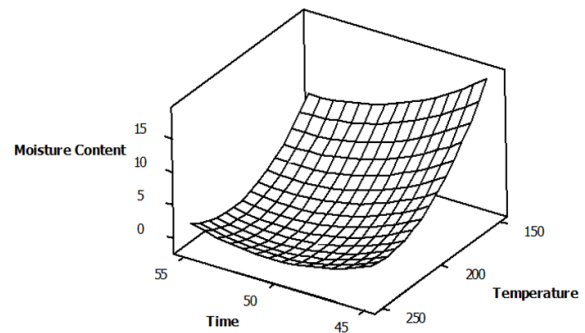


Figure 2a. Interaction effects of time and temperature on moisture content.

In acid number response, temperature factor contributed to negative linear effect and positive non-linear effect whereas time factor showed significant negative linear effect. Figure 2b shows the interaction effects of time and temperature on the acid number. The results indicated that acid number significantly dropped as temperature increased but affected moderately by time. In the first order polynomial, it was found that temperature had a negative effect on lightness. Whilst a positive effect on secondary order polynomial of time was found on lightness. In Figure 2c, lightness was affected by temperature factor and decreasing trend with increased of temperature was observed. Time factor caused smaller changes to lightness which the lightness decreased slightly with increased of drying time. According to Krokida *et al.* (1998), thermal processing affects the food quality either by discoloration or browning which are caused by various reactions, including Maillard reactions, phenol polymerization, and pigment destruction. The lightness of *G. cambogia* fruit rinds decreased with increased of drying time and temperature were mainly attributed to Maillard reaction. The browning reaction of dried fruit rinds was supported by Raju and Reni (2001) which reported *G. cambogia* fruit rinds is rich in non-volatile acid and contains 15% reducing sugar and the complete removal of moisture to a level of 15 to 20% yield a coal black colour dried rinds.

Determination the optimum condition of superheated steam drying to obtain a preferable dried *G. cambogia* fruit rind quality with moisture content target at 15.5%, maximum acid number and lightness shown in Figure 3. The composite desirability of the desired optimum

Surface Plot of Acid Number vs Temperature, Time

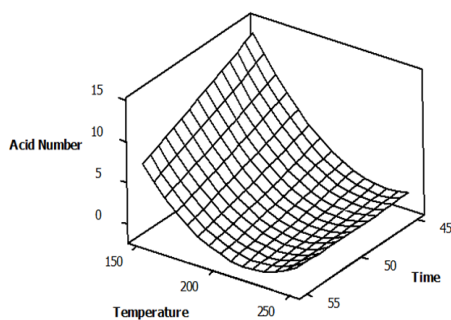


Figure 2b. Interaction effects of time and temperature on acid number.

Surface Plot of Lightness vs Temperature, Time

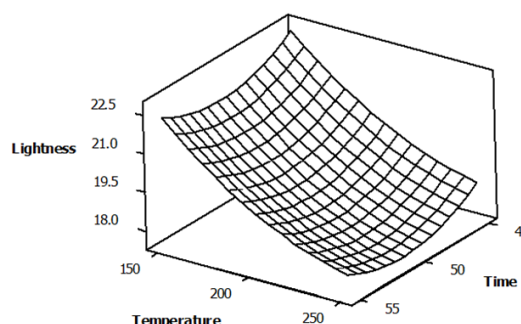


Figure 2c. Interaction effects of time and temperature on lightness.

factors for dried fruit rinds was 0.93375 with a drying time of 46.6508 mins and drying temperature 150°C. In order to make the drying parameters feasible in experimental runs, the drying time is adjusted to 46.6 minutes and drying at 150°C. From the adjustment the composite desirability and individual desirability of moisture content are decreased slightly, however with the individual desirability of acid number and lightness increased slightly. Therefore, the optimum drying time and temperatures of *G. cambogia* is 46.6 mins and 150°C respectively with desirability value 0.91378.

drying time due to the higher drying rates. The acid number of *G. cambogia* was shown affected by temperature and moisture content. The acid number increased in the early stage of superheated drying and decreased when moisture content was significantly reduced. Page model was found to be the best in describing the drying behaviour of *G. cambogia* as compared to Lewis and Henderson-Pabis models. Optimization of superheated steam drying of *Garcinia cambogia* was done by response surface methodology employing central composite design. Time and temperature were the independent variables and found to have significant effects on the dependent variables: moisture content, acid number and lightness. The optimum conditions of drying time and drying temperature of *G. cambogia* were identifies as 46.6 mins and 150°C respectively with desirability value 0.91378. These findings indicate that the kinetic drying of *G. cambogia* by superheated steam could be done to fit into the larger scale of industrial drying.

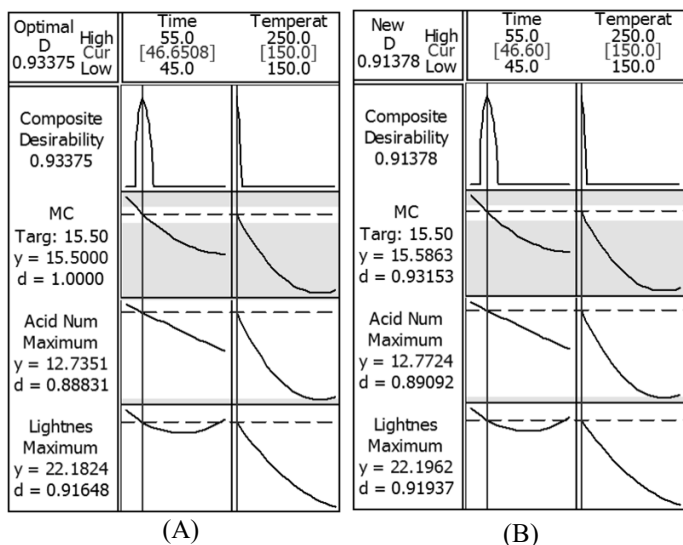


Figure 3. Optimization plot of superheated steam drying of *G. cambogia* with composite desirability and individual desirability of each response.

4. Conclusion

This research studies the kinetic drying of *G. cambogia* by sun drying and superheated steam drying method. In general, kinetic drying of *G. cambogia* by superheated steam drying demonstrated constant drying rate and falling rate. Equilibrium moisture content of *G. cambogia* was achieved within 100 mins by superheated steam drying whilst sun drying required more than 6 hrs. Superheated steam drying provides a significant shorter

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